# 2020 Basic Nuclear Engineering I Lecture note (2)

- Nuclear Fission –

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2. Nuclear Fission

2.1 Nuclear fission by neutron

•Heavy nucleus causes fission reaction by the collision with a neutron with

very high speed

 $\rightarrow$  =with large <u>mass number</u>

 $\rightarrow$  Number of protons + neutrons

·Heavy nucleus with odd number of neutrons causes fission reaction by

very slow neutrons

•Nucleus which satisfies the condition in nature is uranium 235 only

(<sup>235</sup>U)

Mass number 235 -92 Atomic number

 $\begin{bmatrix} \text{Number of protons: } 92 \\ \text{Number of neutrons: } 235 - 92 = 143 \end{bmatrix}$ 

•Example of nuclear fission of <sup>235</sup>U <sup>235</sup><sub>92</sub> U +  ${}^{1}_{0}$  n  $\rightarrow {}^{236}_{92}$  U  $\rightarrow {}^{141}_{56}$  Ba +  ${}^{92}_{36}$  Kr + 3  ${}^{1}_{0}$  n

Uranium235 neutron

Uranium236 (unstable) Barium141 + Kripton92 + three neutrons

Various combinations

two or three neutrons emission

Mass defect in the reaction

 $\simeq 3.6 \times 10^{-28} \text{ kg}$ 

Corresponding energy (by  $E = mc^2$ )

 $\simeq 3.2 \times 10^{-11} \text{ J}$  ( $\cong 200 \text{ MeV}$ )

Ref. In the chemical reaction

 $C+O_2 \ \rightarrow \ CO_2$ 

Release energy  $\simeq 7 \times 10^{-19} \text{ J}$ 

 $(3.2 \times 10^{-11}) / (7 \times 10^{-19}) \simeq 4.6 \times 10^{7}$ 

 $\cdot$ Number of neutrons emitted in fission reaction v

 $v \simeq 2.5$  (in case of <sup>235</sup>U)

• Energy of neutrons emitted in fission reaction (fission neutron)

average 2MeV ( $1eV = 1.602 \times 10^{-19}J$ ) peak 1MeV

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### 2.2 Actinoid

 $\circ$ Actinoid : element whose atomic number is from <u>89</u> to <u>103</u>

Ac Lr

oImportant actinoid in nuclear reactor

•Uranium (Atomic number 92)

Abundance in the earth's crust :  $2 \times 10^{-6}$ 

Natural uranium 235  $^{235}$ U 0.71% fissile  $\leftarrow$  easy to cause fissile by neutron uranium 238  $^{238}$ U 99.29%

•Thorium (Atomic number 90)

Abundance in the earth's crust :  $7 \times 10^{-6}$ 

Natural thorium ···· thorium 232 <sup>232</sup>Th 100% non-fissile

<sup>232</sup>Th : by neutron capture

transmuted to 233U (fissile)

does not exist in nature

•Plutonium (Atomic number 94)

does not exist in nature

<sup>238</sup>U : by neutron capture

transmuted to <sup>239</sup>Pu (fissile)

3. Nuclear fission chain reaction

## 3.1 Concept of nuclear fission chain reaction

By using neutrons emitted by fission reaction, causing the next fission

reaction

neutron + <sup>235</sup>U  $\xrightarrow{\text{fission}}$  neutrons + energy + light nuclei  $(2 \sim 3)$ But !!

Even if neutrons are injected to natural uranium, fission chain reaction does not occur.

(3.1. to be continued)