

Structural Steels and Their Weldability

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Chemical Composition of Steels
Weldability
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Lamellar Tearing

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Historical Review of Steels

Iron and Steel

Iron

Fe Ductile Material, Tensile Strength=200-300MPa

Steel

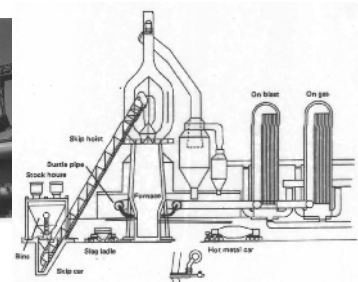
Removal of Impurities

Fe + Additional Alloy Metals
(C+Mn+Si+Cr+...)

Improved Properties

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Production of Iron



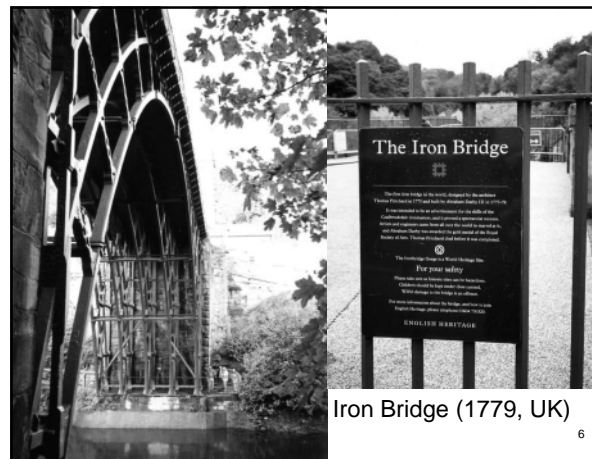
Pig Iron and Slag

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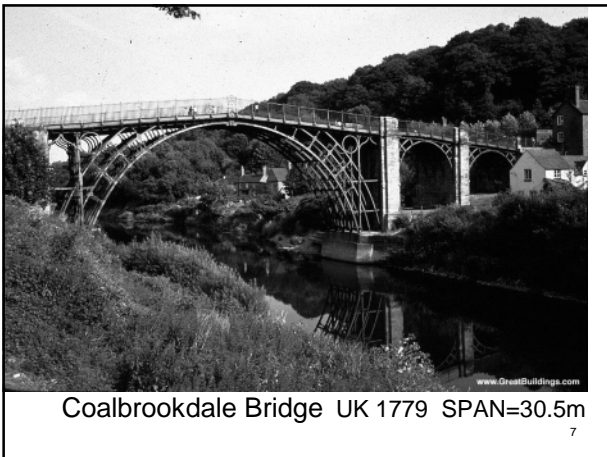
Iron Bridge (1779, UK)

The First Iron Bridges



Iron Bridge (1779, UK)

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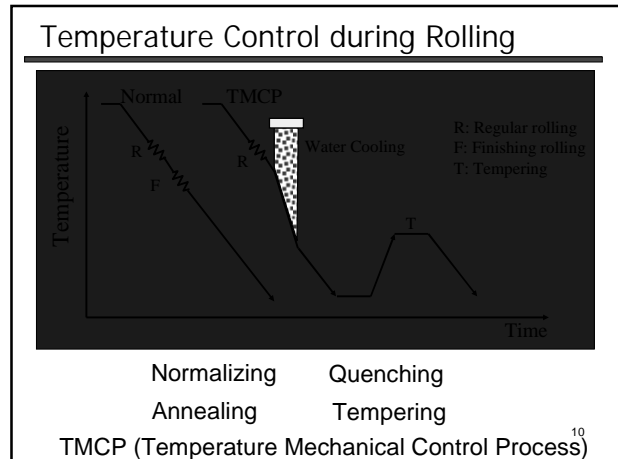
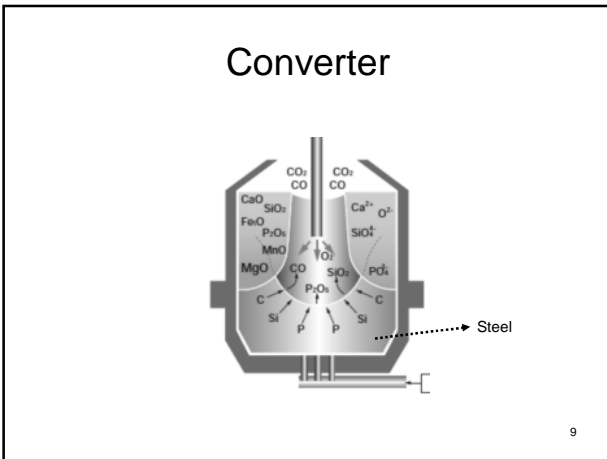


Production of Steels

Impurities in pig iron are removed by oxidation in a vessel called a converter

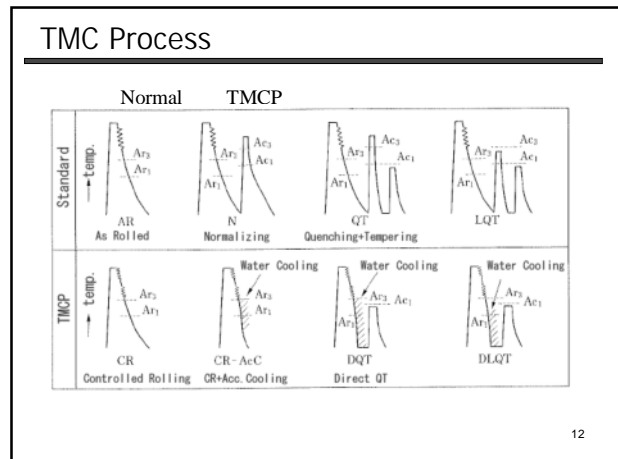
Bessemer Process

Pig Iron C:2.5-4.5%
→ Cast Iron

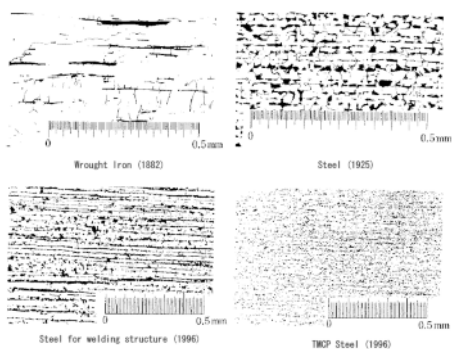


Temperature Control during Rolling

- Normalizing: Heat up to austenite temperature ($A_3 + 20-30$ degree) keep it for a while, cool down in the air. Finer and even crystallization, improve ductility and toughness
- Annealing: Heat up products to appropriate temperature and cool down slowly. To soften steel, release residual stress
- Quenching: Heat steel to A_3 or $A_1 + 30-50^\circ$ keep it for a while, and cool down quickly, Hardness of steel is dependent on quantity of martensite
- Tempering: Improve the property of steel after quenching by heating the steel below A_1 point. QT steel is processed in quenching and tempering.
- TMCP (Temperature Mechanical Control Process)
Cooling speed is controlled during rolling. Fine crystal, higher strength with lower carbon equivalent content. (good weldability) It is possible to lower the yielding ratio (ratio of yield strength to tensile strength)



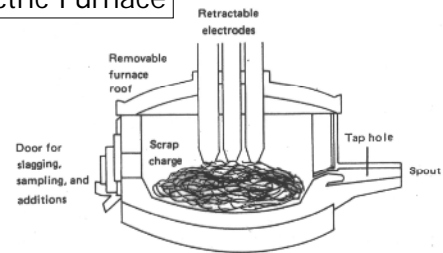
Granular size



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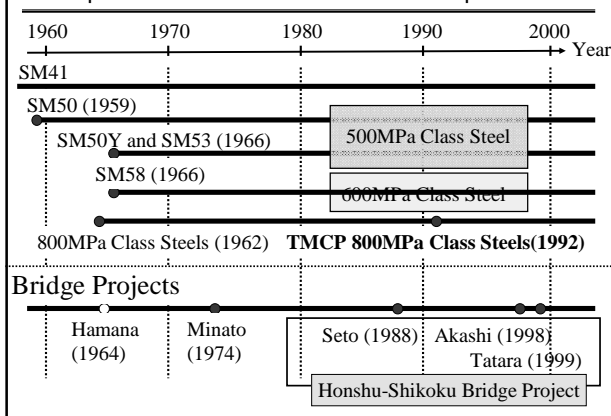
Recycling Steels from Scrap

Electric Furnace



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Development of Structural Steels in Japan



Effects of Impurities and Alloy Metals

Chemical Composition Example: JIS-SM490YB, SM570Q

	C	Si	Mn	P	S	Cu	Ni	Cr	V
SM490YB	0.14	0.46	1.56	0.020	0.005	0.01	0.01	0.02	0.04
SM570Q	0.14	0.23	1.44	0.012	0.005				

Carbon, C Very Influential Element

Carbon acts as Hardener and Strengthener, but reduces the Ductility

Silicon, Si

Less 0.2%: Slight Effects on Strength and Ductility
0.3-0.4%: Elastic Limit and Strength are Raised

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Effects of Impurities and Alloy Metals

Nickel, Ni

Nickel-Chromium Steels have a very high tensile strength with considerable toughness and ductility, improves corrosion resistance

Stainless Steels

Chromium, Cr

Same as Ni, hardening Agent, Intense Hardness after quenching with very high strength
Chromium improves corrosion resistance

Copper, Cu

Increases the resistance of steels to atmospheric corrosion
Weathering Steels
increase strength and hardness but decrease ductility

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Effects of Impurities and Alloy Metals

Sulfur, S

Less than 0.1% : No appreciable effects
It has a very injurious effects upon the hot metal, Lessening its malleability and weldability

Phosphorus, P

The Most undesirable impurity
It decrease toughness, ductility, and weldability
It Improves Corrosion Resistance

Manganese, Mn

Improves the strength of carbon steel, decrease ductility

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Mechanical Properties of Steels

Governed by Chemical Composition

	Mn	Si	P	S	Ni	Cr	Cu
Strength							
Ductility	x						x
Toughness			x				
Hardness							
Corrosion Resistance							
Weldability			x	x			

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Steels

Iron



Steel

Chemical Composition

Improvement of Mechanical Properties
High Strength
High Ductility
Mass Production
High Quality

Steels:

Tensile Strength: 400, 500, ..., 800MPa

Cables:

..., 1100, ..., 1600, 1800 MPa

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Weld Cracks

Hot Cracks

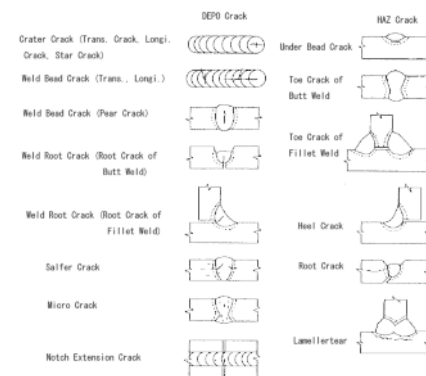
Cold Cracks

Lamellar Tearing



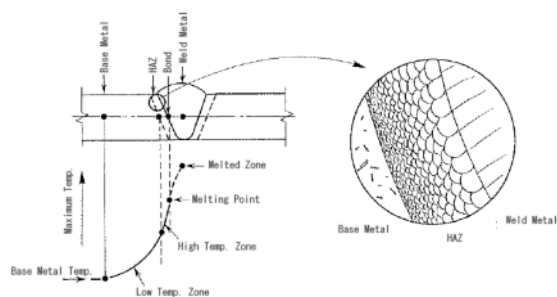
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Weld Cracks



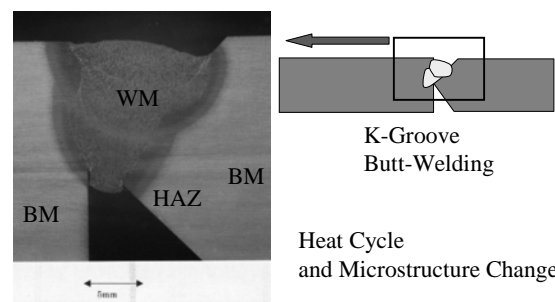
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Micro structure of weld metal



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HAZ in Welded Joints



Heat Cycle
and Microstructure Change

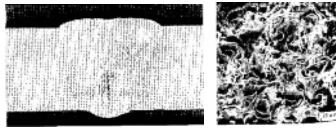
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Hot Cracks

Cracks occur near melting temperature during welding or after welding

Hot Cracks can be eliminated by designing the material composition of weld metal and base metal.

$$HCS = \frac{C(S + P + \frac{Si}{25} + \frac{Ni}{100})}{3Mn + Cr + Mo + V} \times 10^3 < 4$$



Cracks occurred during solidification

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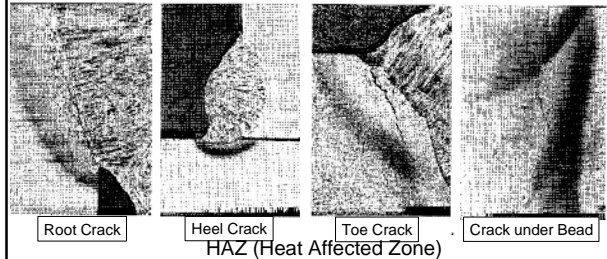
Cold Cracks

Generally, Tran-granular cracks

Short time cracking: during cooling process

Delayed cracking: after some time lapse at room temperature

Hydrogen is responsible for cold cracks



HAZ (Heat Affected Zone)

Resistance Performance of HAZ against Cold Crack

Carbon Equivalent

$$C_{eq} = C + \frac{Mn}{6} + \frac{Si}{24} + \frac{Ni}{40} + \frac{Cr}{5} + \frac{Mo}{4} + \frac{V}{14} \left(+ \frac{Cu}{13} \right)$$

$$P_{CM} = C + \frac{Si}{30} + \frac{Mn}{20} + \frac{Cu}{20} + \frac{Ni}{60} + \frac{Cr}{20} + \frac{Mo}{15} + \frac{V}{10} + 5B$$

IIW

$$CE_{IIW} = C + \frac{Mn}{6} + \frac{Cu + Ni}{15} + \frac{Cr + Mo + V}{5}$$

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Prevention of Cold Cracks

Use of Low C_{eq} or P_{CM} steel

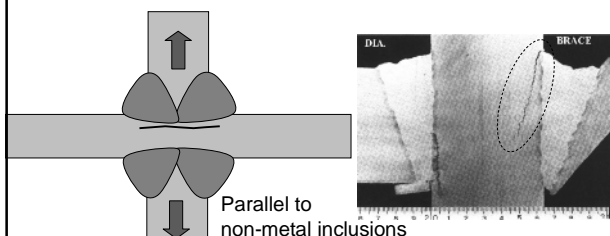
Use of Low Hydrogen Processes

Preheating

Reduce Joint Restraint

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Lamellar Tearing



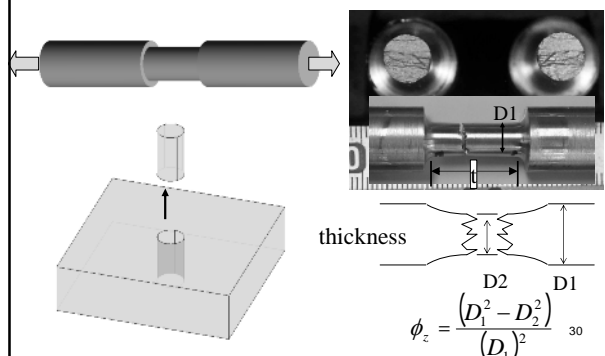
Parallel to non-metal inclusions

High Constraint Condition
Large Deposited Weld
Ductility in Plate Thickness Direction

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Z-tension tests

- z: Reduction of Area in Thickness Direction (Z-Direction) Tension tests



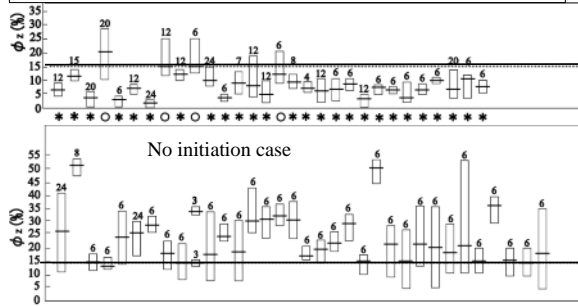
thickness

$$\phi_z = \frac{D_1^2 - D_2^2}{(D_1)^2} \quad 30$$

Lamellar tearing and ϕ_z

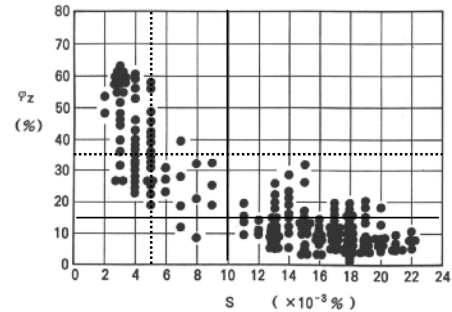
- $\phi_z < 15\%$, High risk of Lamellar tearing

Crack initiating cases of real fabrication



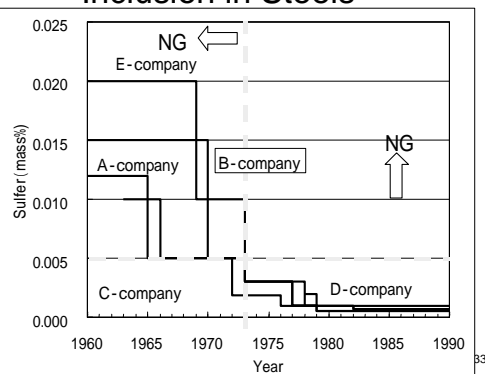
ϕ_z and Sulfur Inclusion in Steels

- If $S > 0.01\%$ then many of the ϕ_z -tension test results are $\phi_z < 15\%$.



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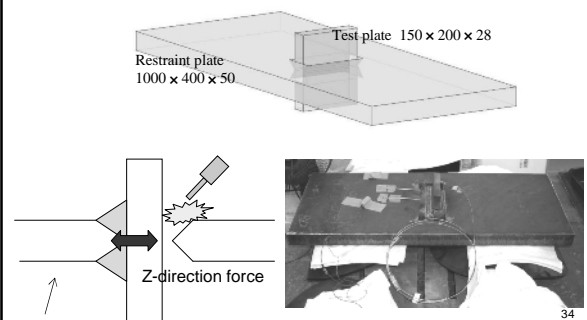
Historical Review of Sulfur Inclusion in Steels



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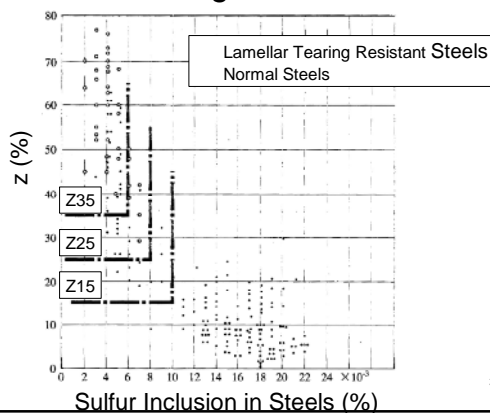
Z-Window Welding Tests

To Evaluate the possibility of Lamellar Tearing



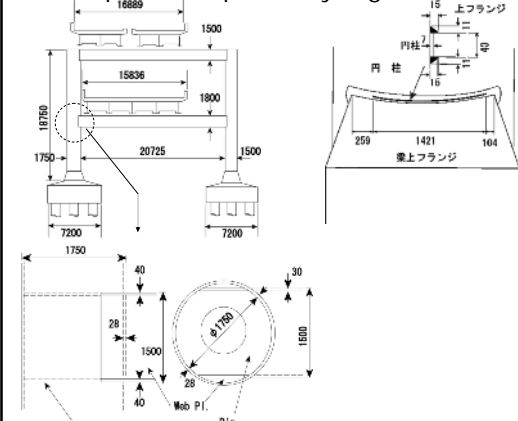
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Lamellar Tearing Resistant Steels



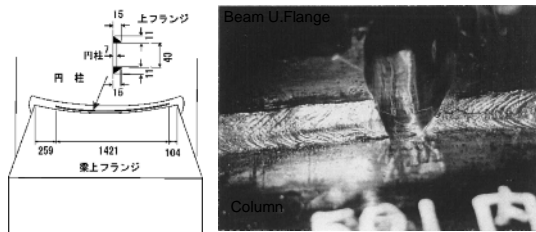
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Metropolitan Expressway Higashi Ikebukuro



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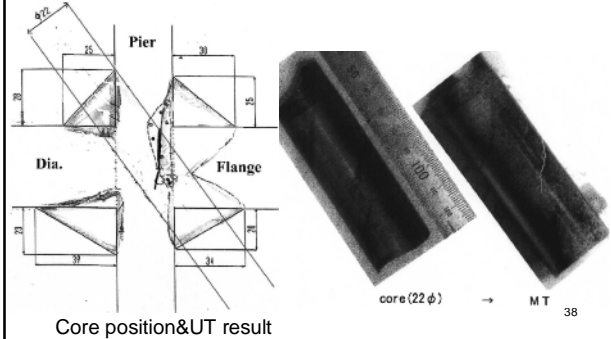
Fatigue Crack (Inside of the Pier)



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Material property(1)

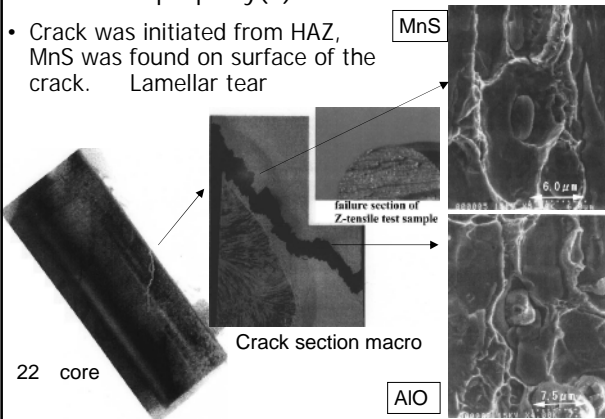
- To check the result of UT, a core sample was taken from welded section. Crack due to lamellar tearing was found in the pier column.



Core position&UT result

Material property(2)

- Crack was initiated from HAZ, MnS was found on surface of the crack. Lamellar tear



22 core

Crack section macro

AIO

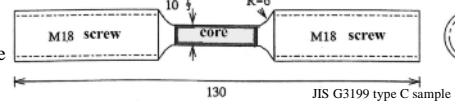
Material property(3)

- Sulfur level is not so high, compared to contemporary steels
- Some samples show very low RAZ.

Chemical compositions													
sample No.	C	Si	Mn	P	S	Ni	Cu	Cr	Mo	V	Ceq	Pcm	parameters
No.1	0.16	0.29	1.36	0.012	0.012	0.017	0.04	0.018	0.012	0.003	0.406	0.242	
No.2	0.17	0.29	1.36	0.012	0.01	0.017	0.04	0.018	0.012	0.003	0.416	0.252	
No.3	0.17	0.29	1.35	0.012	0.009	0.016	0.04	0.017	0.012	0.003	0.414	0.251	

Z-direction tensile test result									
sample No.	diameter (mm)	0.2% load Load(kN)	tensile stress N/mm ²	Load(kN)	tensile stress N/mm ²	elongation (%)	RAZ (%)	break point connection	
No.1a	9.99	26.81	342	27.89	356	3	0.2	A	
No.1b	9.99	28.34	362	39.78	508	12	8	A	
No.1c	9.99	27.57	352	40.13	512	10	12	A	
No.2a	10.00	26.49	337	39.41	502	18	20	A	
No.2b	10.00	26.39	336	40.16	511	10	19	A	
No.2c	9.99	24.92	318	36.84	470	6	10	A	

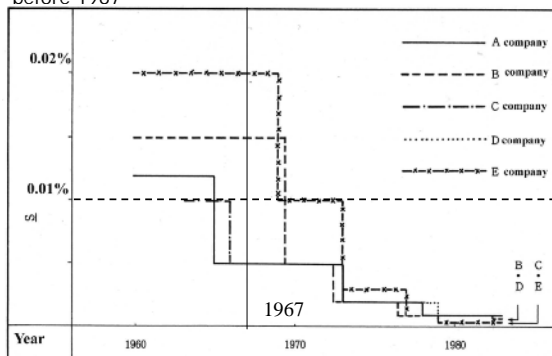
Bolt hole core sample



JIS G3199 type C sample

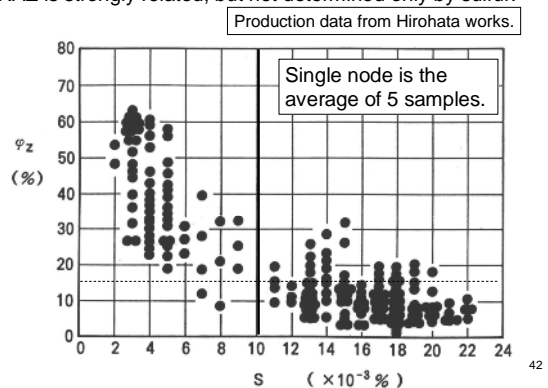
History of Sulfur Inclusion in Steel

- Difficult to Apply Weld Repairing to Piers constructed before 1967

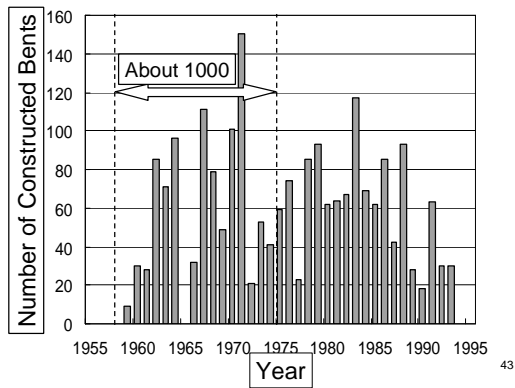


Sulfur and RAZ

- RAZ is strongly related, but not determined only by sulfur.



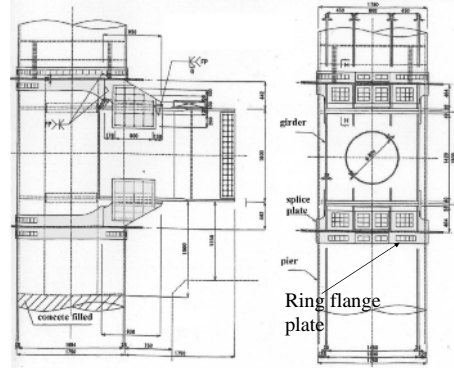
Construction Year and Number of Constructed Steel Bents in MEX



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Support members for welding

- Decreasing the traffic stress by more than 50%.



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